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## Reaction of Allenylphosphonic Diamides with Bromine

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## REACTION OF ALLENYLPHOSPHONIC DIAMIDES WITH BROMINE

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In contrast to allenylphosphonic dichlorides and dial-kyl esters, the diamides are little known and practically not studied. The allenylphosphonic diamides <u>2a-h</u> were prepared from the 3,3-disubstituted allenylphosphonic dichlorides <u>1a,b</u>,  $CI_2P(0)-CH=C=CR^1R^2$ , and heterocyclic amines at law temperature. Here we report the results of the reaction of <u>2a-h</u> with bromine. Unlike the halogenation of the corresponding dialkyl esters, dichlorides and tertiary phosphine oxides, where only one reaction route has been observed-cycloaddition or 2,3-addition, both reaction routes proceeds in allied degree in the case studied:

$$R^{3}N$$
  $P-CH=C=C$   $R^{2}$   $+$   $Br_{2}$   $-10-0^{\circ}C$   $R^{3}N$   $P-CH=C=C$   $R^{2}$   $+$   $R^{2}$   $+$   $R^{3}N$   $P-CH=C$   $R^{3}N$   $R^{4}N$   $R^$ 

The reaction products ratio depends on the R<sup>3</sup>N-rest, the degree of heterocyclization being higher in all cases. The formed phosphonium salts <u>3a-h</u> were isolated as crystalls and appears to be stable substances. The structures of <u>2a-h</u>, <u>3a-h</u> and <u>4a-h</u> were confirmed by IR, MS, <sup>1</sup>H- and <sup>13</sup>C NMR. The high chemo- and Z-stereoselectivity of the reaction is discussed.